TASK 1: ALL Observations and data should be reported on the answer sheet!

- 1. Obtain a sample of solid CuCl₂ from the supervisor, and using a beaker, collect a sample of at least 50mL of water. Record observations about what these things look like–include temperature and conductivity of the water
- 2. Mix the solid into the water, stir well.
- 3. Once the solid has dissolved, record temperature, conductivity and any other observations.
- 4. Keep the beaker and its contents for Task 2
- 5. Answer the following questions ON YOUR ANSWER page about task 1:
 - a. Was there a change in temperature, if so, how much did it change?
 - b. Was there a change in the conductivity?
 - c. What kind of compound is copper II chloride (CuCl₂)?
 - d. What evidence do you have to support your answer to c?
 - e. In TASK1, what type of change occurred? Justify your answer.
 - f. Suggest a method that could be used to separate the substance at the end back into the 2 substances you mixed.

TASK 2: ALL Observations and data should be reported on the answer sheet!

- 1. Collect a piece of aluminum foil from the supervisor. Make observations
- 2. Take the temperature of the contents in the beaker from Task 1.
- 3. Stir the aluminum foil into the contents of the beaker from Task 1.
- 4. Make observations as you are stirring, does the temperature change?
- 5. Once you feel that the change has stopped, take a final temperature and make observations.
- 6. Answer the following questions ON YOUR ANSWER page about task2:
 - a. What kind of change happened? Justify your answer
 - b. Did the temperature change? If yes, was the change exothermic or endothermic?
 - c. Which diagram (on your answer sheet)best represents what you saw?
 - d.Write a balanced equation that represents the change you saw. (tie breaker)

CLEANUP- You will pour the contents of your beaker into the waste container indicated by your supervisor. Do not pour anything down the drain. Make sure you wash your hands!